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# Contributions to the chemistry of Mo and W. XVIII.\* The structures of the triclinic and orthorhombic

**modifications of WO<sub>3</sub>.** By F. A. SCHRÖDER, Gmelin-Institut für Anorganische Chemie der Max-Planck-Gesellschaft, D-6 Frankfurt/M. 90, Varrentrappstr. 40/42, Deutschland (BRD)

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The structure of the triclinic modification of WO<sub>3</sub> in terms of the  $O \rightarrow W$  electric dipoles is proposed. In the same way, based on the relationships with the structures of MoO<sub>3</sub> and WO<sub>2</sub>Cl<sub>2</sub>, the structure of the orthorhombic modification of WO<sub>3</sub> is proposed.

## (a) Triclinic WO<sub>3</sub>

Van Landuyt & Amelinckx (1969) first showed that the model of electric dipoles for the long  $O \rightarrow W$  bonds is used very successfully to explain the well known (100) domains in monoclinic WO<sub>3</sub>. Schröder & Hartman (1972) discussed the problem thoroughly, considering all existing chains of electric dipoles (along [100] and [001]). The result is that in the structure section  $d_{002}$  of the monoclinic unit cell both types of chains (two from each) are arranged corresponding to a minimum of repulsion that is mutually antiparallel. The second section  $d_{002}$  is (because there are two independent formula units in monoclinic WO<sub>3</sub>) quasi-translational with respect to the first one. This leads to a total of two parallel [010] dipole chains on the common (002) face. This boundary repulsion is obviously low enough for the unit cell to exist while a total of four parallel dipole chains (two [010] and two [001]) at the (100) boundary leads to the (100) domain growth [cf. Fig. 5 in Schröder & Hartman (1972)]. Schröder (1973) postulated the existence of another WO<sub>3</sub> unit cell of nearly the same size, which is derived formally from the known form by rotation of the [010] chains (this means the WO<sub>3</sub> octahedra) by 180° around the [001] dipole chain 'axis'. This unit cell is shown in Fig. 1 and is here attributed to one of the suggested low-temperature modifications of WO:

Taking into account the very recent investigations on this subject [Salje, 1974; Salje & Viswanathan, 1975; Salje, 1975, hereafter referred to as S. (1974), S. & V. (1975) and S. (1975)] it is supposed that this derived unit cell is that of the triclinic structure, WO<sub>3</sub>(II) according to S. & V. (1975). This is supported by the following points:

(1) Because of the lack of internal repulsion from parallel dipole chains, the volume of the derived unit cell should be smaller than that of the monoclinic unit cell. This has been confirmed: 421.8 (2) instead of 422.8 (2) Å<sup>3</sup> (S. & V., 1975). The volume has been doubled for both cells since from experience of X-ray powder data of monoclinic WO<sub>3</sub>. it is known that C=c/2 is always obtained by this method.

(2) The derived triclinic cell would be space group P1 which allows ferroelectric or antiferroelectric behaviour, observed several times, contrary to the known centrosymmetric monoclinic space group.

(3) The only noticeable difference between the lattice parameters of monoclinic and triclinic WO<sub>3</sub> is found to be a contraction in *b*. It may be concluded that the short W-O<sub>y</sub> ( $\pi$ -bonding) distances are shortened from an average of 1.763 Å to approach the W-O<sub>z</sub> distances (av. 1.736 Å).

## (b) Orthorhombic WO<sub>3</sub>

Schröder & Felser (1972) described crystals of monoclinic WO<sub>3</sub> showing (100) domains. The X-ray pattern showed reflexions hk0: k = 2n + 1 only, after long exposures of Weissenberg photographs. Later Schröder (1973), from this orthorhombic pseudosymmetry, concluded that the space group of the orthorhombic WO<sub>3</sub> should be Pmnb (Standard: Pnma). The same result was obtained recently (S. & V., 1975). This is the same space group shown by the usual orthorhombic structure of MoO<sub>3</sub>, although this structure was solved in an alternative orientation, namely Pbnm [Last refinement Kihlborg (1963/4)]. From this information the structure of orthorhombic WO<sub>3</sub> may be predicted and demonstrate the influence of dipole chains in systemizing the modifications of WO<sub>3</sub>. Table 1 shows the hitherto measured lattice parameters of orthorhombic WO<sub>3</sub>. The most recent values show satisfactory agreement. The lattice parameters of MoO<sub>3</sub> according to Kihlborg (1963/4), in the orientation *Pmnb* are included in Table 1. It is seen that  $2a (MoO_3)$  $\approx a$  (o'rh.WO<sub>3</sub>),  $b(MoO_3) \sim 2b$  (o'rh.WO<sub>3</sub>) and  $c(MoO_3) \approx$  $c(WO_3)$ . While the agreements for a and c are satisfactory. there is a difference of 7.570-6.927-0.643 Å in b.



Fig. 1. The proposed unit cell of triclinic WO<sub>3</sub>. The described rotation has been carried out on the  $d_{002}$  section in the front of the figure.

<sup>\*</sup> For part XVII see Z. Naturforsch. (1975). B30, 638.

Fig 2 shows the structure of orthorhombic WO<sub>3</sub> by the usual idealized octahedra. The axes are given according to space group *Pmcn* (a further alternative orientation of *Pnma*) by which the structure is described most conveniently. The structure of orthorhombic WO<sub>3</sub> is obtained from that of MoO<sub>3</sub> formally by translating every second double layer by  $\pm a/2(MoO_3, Pbnm) = \pm c/2(Pmnb)$ . The resulting arrangement is one of almost c.c.p. of the O atoms. A certain extension along [010] (MoO<sub>3</sub>) is expressed by the above calculated difference of ~0.6 Å. In fact, the deviation is small as the ratio volume of the unit cell/number of O atoms shows: triclinic WO<sub>3</sub> 17.58, monoclinic WO<sub>3</sub> 17.98 Å<sup>3</sup>. The value for MoO<sub>3</sub> is 16.9 Å<sup>3</sup>.

The most general among the fourfold equipoints in this space group No.  $62 - D_{2h}^{26}$  is 4(c). If this is occupied by the atoms of WO<sub>3</sub>, the W atoms should have coordinates close to 000,  $0\frac{1}{2}0$ ,  $\frac{1}{2}0\frac{1}{2}$ ,  $\frac{1}{2}\frac{1}{2}$  as can be concluded from Fig. 2. If



Fig. 2. The proposed structure of orthorhombic  $WO_3$  drawn as idealized octahedra. The crystal axes correspond to space group *Pmcn*.

the W atoms do have these coordinates exactly [which represent equipoint 4(a)] greater differences in bond distances as compared with those in MoO<sub>3</sub> must result. This is most likely so, as it is in accord with the differences in structural behaviour between Mo and W. How far pseudosymmetry in orthorhombic WO<sub>3</sub> towards the space group *Bbmm* (standard *Cmcm*; all atoms on *mm*), found in Mo<sub>4</sub>O<sub>10</sub>(OH)<sub>2</sub>, will be realized, can be seen only by the structure determination. The lattice parameters of this compound are given also in Table 1. The structure of orthorhombic WO<sub>3</sub> already contains the arrangement of two kryptocells of tetragonal WO<sub>3</sub> (Kehl, Hay & Wahl, 1952).

The true arrangement of the electric dipole chains in orthorhombic  $WO_3$  can of course be elucidated only by the structure determination, but by analogy with  $MOO_3$ , so far no reasons for internal repulsions can be seen.

To a certain degree, the arrangement of dipole chains in orthorhombic WO<sub>3</sub> can be predicted by comparison with those in WO<sub>2</sub>Cl<sub>2</sub> (Jarchow, Schröder & Schulz, 1968).\* This structure also has W-O dipole chains in two directions, namely along [001] and [010] (space group B11b). The first ones are antiparallel (two per unit cell) while the second are parallel by translation. Structural equivalence of lattice directions between WO<sub>2</sub>Cl<sub>2</sub> (A11a) and WO<sub>3</sub> (P2<sub>1</sub>/n) has been shown (Schröder, 1973) to exist in the following way  $[100] \triangle [010]$  and  $[010] \triangle [001]$ . If the setting B11b is used, corresponding directions in  $WO_2Cl_2$ ,  $WO_3$  ( $P2_1/n$ ) and  $WO_3$ (Pmnb) are along equivalent lattice directions, neglecting the obliqueness of the WO<sub>2</sub>Cl<sub>2</sub> cell and, because of the centring of this cell, taking only half its corresponding lattice parameter. From comparison with MoO<sub>3</sub> (Pmnb) where we have dipoles along [010] and [001], we may assume that in orthorhombic WO<sub>3</sub> (Pmnb) they are also along these directions.

\* In the paper of Jarchow, Schröder & Schulz (1968), the labelling of the atoms in Fig. 4 is incorrect. W(1) is linked to Cl(1), Cl(2), O(1), O(2) with the coordinates given in Table 2 [except those for Cl(1) and Cl(2)] and to two more symmetryrelated O atoms, labelled O(3) and erroneously O(2) in Fig. 4. The coordinates of Table 2 refer to the given orthorhombic cell. The transformation matrix applies only to the z values of the Cl atoms. The equivalent positions for A1a1 are: x, y, z;  $\frac{1}{2} + x, \overline{y}, z; x, \frac{1}{2} + y, \frac{1}{2} + z; \frac{1}{2} + x, \frac{1}{2} - y, \frac{1}{2} + z; c = 14.41$  Å.

	а	Ь	С	Space group	Reference
WO <sub>3</sub>	7.13	7.33	3.84		(i)
$WO_{3}$ (>350°C)	7.28	7.45	3.81		(ii)
WO <sup>*</sup>	7.339	7.545	3.857		(iii)
WO3	7.336	7.560	3.872		(iv)
WO <sub>3</sub>	7-356	7.583	3.878		(v)
WO3	7.341 (4)	7.570 (4)	3.877 (2)	Pmnb	(vi)
5	3.877	7.341	7.570	Pmcn	
WO <sub>3</sub> (RT)	7.306 (1)	7.540(1)	7.692 (1)	$P2_1/n$	(vii)
$WO_2Cl_2(RT)$	7.68 (1)	3.89 (1)	14.41 (2)	Alal	(viii)
	7.20†	7.68	3.89	B11b	
MoO <sub>3</sub> (RT)	3.6964 (6)	13.855 (3)	3.9628 (7)	Pmnb	(ix)
$Mo_4O_{10}(OH)_2(RT)$	3.734	14.082	3.888	Bbmm	(x)

Table 1. Lattice parameters (Å) and space groups for WO<sub>3</sub>, WO<sub>2</sub>Cl<sub>2</sub>, MoO<sub>3</sub> and Mo<sub>4</sub>O<sub>10</sub>(OH)<sub>2</sub>

References: (i) Ueda & Ichinokawa (1951), (ii) Wyart & Foex (1951), (iii) Rosen, Banks & Post (1956), (iv) Perri, Banks & Post (1957), (v) Ackermann & Sorrell (1970), (vi) Salje & Viswanathan (1975), (vii) Loopstra & Rietveld (1969), (viii) Jarchow, Schröder & Schulz (1968), (ix) Kihlborg (1963/64), (x) Wilhelmi (1969).

\* Lattice parameters were taken from diagrams for 480°C; † the value is 14.41/2 Å, the original lattice parameter of 13.90 Å for the monoclinic cell is incorrect (cf. footnote on this page); RT: room temperature value.

Some comments on the papers S. (1974), S. & V. (1975) and S. (1975) are given in the Appendix.\*

Note added in proof: The most recently published results on the doubling of the *c* axis in monoclinic WO<sub>3</sub> (Lefkowitz, Dowell & Shieldt, 1975) confirm the equivalence of the lattice directions in WO<sub>2</sub>Cl<sub>2</sub>, monoclinic WO<sub>3</sub> and orthorhombic WO<sub>3</sub> given here, as well as earlier results (Schröder & Felser, 1972).

#### References

- ACKERMANN, R. J. & SORRELL, C. A. (1970). *High Temp. Sci.* **2**, 119–130.
- JARCHOW, O., SCHRÖDER, F. & SCHULTZ, H. (1968). Z. anorg. allgem. Chem. 363, 58-72.
- Kehl, W. L., HAY, R. G. & WAHL, D. (1952). J. Appl. Phys. 23, 212–215.

KIHLBORG, L. (1963/4). Ark. Kem. 21, 357-364.

\* The Appendix has been deposited with the British Library Lending Division as Supplementary Publication No. SUP 31508 (4 pp.). Copies may be obtained through The Executive Secretary, International Union of Crystallography, 13 White Friars, Chester CH1 1NZ, England.

- LEFKOWITZ, I, DOWELL, M. B. & SHIELDS, M. B. (1975). J. Solid State Chem. 15, 24–39.
- PERRI, J. A., BANKS, E. & POST, B. (1957). J. Appl. Phys. 28, 1272–1275.
- ROSEN, C., BANKS, E. & POST, B. (1956). Acta Cryst. 9, 475-477.
- SALJE, E. (1974). J. Appl. Cryst. 7, 615-617.
- SALJE, E. (1975). Acta Cryst. A31, 360-363.
- SALJE, E. & VISWANATHAN, K. (1975). Acta Cryst. A31, 356–359.
- SCHRÖDER, F. A. (1973). Habilitationsschrift, Freiburg i. Br., p. 40ff.
- SCHRÖDER, F. A. & FELSER, H. (1972). Z. Kristallogr. 135, 391-398.
- SCHRÖDER, F. A. & HARTMAN, P. (1972). Z. Naturforsch. 27b, 902–908.
- UEDA, R. & ICHINOKAWA, T. (1951). J. Phys. Soc. Japan 6, 122–123.
- VAN LANDUYT, J. & AMELINCKX S. (1969). Cryst. Lattice Defects, 1, 113–116.
- WILHELMI, K. A. (1969). Acta Chem. Scand. 23, 419-428.
- WYART, J. & FOEX, M. (1951). C.R. Acad. Sci. Paris 233, 2459-2461.

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**The residual in the case of a mixture of centric and acentric reflexions.** By P. E. NIXON\*, Laboratory of Molecular Biophysics, South Parks Road, Oxford OX1 3PS, England

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The value of the residual for a randomly wrong structure when calculated from a mixture of centrosymmetrical and non-centrosymmetrical structure factors is close to the value obtained by interpolation between the centric and acentric cases of Wilson [Acta Cryst. (1950). 3, 397–398].

Wilson (1950) pointed out that it was useful to know the likely value of the crystallographic residual for a randomly wrong structure, and he showed that this was 2/2-2 if the Wilson (1949) centric distribution function applies, and  $2-\sqrt{2}$  if his acentric distribution function applies. In non-centrosymmetrical space groups other than P1, one has both centrosymmetrical and non-centrosymmetrical structure factors, and usually calculates the residual without separation of the structure factors into two sets.

From equations (1) and (5) of Wilson's 1950 paper

where

Rrandom

$$R_{\text{random}} = \frac{2\langle F \rangle - 4\langle G(F) \rangle}{\langle F \rangle}$$
$$G(F) = \int_0^F F' P(F') dF'.$$

If a proportion c of the structure factors belongs to the centric distribution, and (1-c) to the acentric,

$$=\frac{2c\langle F\rangle_{c}+2(1-c)\langle F\rangle_{A}-4c\langle G\rangle_{c}-4(1-c)\langle G\rangle_{A}}{c\langle F\rangle_{c}+(1-c)\langle F\rangle_{A}}$$
(1)

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where the subscripts C and A refer to the centric and acentric distributions used to average F and G. Values of these averages are given by Wilson (1950): substitution in (1) gives

$$R_{\text{random}} = \frac{(4 - 2\sqrt{2})c + \pi(1 - \sqrt{2}/2)}{\sqrt{2}c + \frac{\pi}{2}(1 - c)}$$
(2)

which reduces to Wilson's equations (8) and (10) when c=1and 0 respectively. A straight line interpolation between the limiting values is given by

$$R_{\text{random}} \approx 2 - \sqrt{2} + (3\sqrt{2} - 4)c \approx 0.59 + 0.24c .$$
(3)

Use of (3) rather than the more exact (2) involves an error never greater than 0.006.

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#### References

WILSON, A. J. C. (1949). Acta Cryst. 2, 318–321. WILSON, A. J. C. (1950). Acta Cryst. 3, 397–398.